



Mass Number	Atomic Number	Isotope Notation	Number of Neutrons	Bohr Notation	Electron Dot
31	15	$^{31}_{15}\text{P}$	16	2-8-5	<del><math>\text{P}</math></del>
84	36	$^{84}_{36}\text{Kr}$	48	2-8-18-8	$\text{Kr}$
1	1	$^1_1\text{H}$	0	1	$\text{H}$
14	6	$^{14}_6\text{C}$	8	2-4	$\text{C}$
40	21	$^{40}_{21}\text{Sc}$	19	2-8-9-2	<del><math>\text{Sc}</math></del>
137	56	$^{137}_{56}\text{Ba}$	81	2-8-18-18-8-2	$\text{Ba}$
7	3	$^7_3\text{Li}$	4	2-1	$\text{Li}$

$\text{P}$   
 $\text{Sc}$